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# Two new compounds from Helichrysum arenarium (L.)

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Two new compounds were isolated from the whole plant of *Helichrysum arenarium* (L.) Moench. By means of spectroscopic data (IR, UV, 1D and 2D NMR, HR-MS, ESI-MS, and NOESY) and chemical evidence, the structures were established as 6,7-dimethoxy-4-hydroxy-1-naphthoic acid (1) and (*Z*)-5-hydroxy-7-methoxy-4-[3-methyl-4-( $O-\beta$ -D-xylopyr-anosyl)but-2-enyl]isobenzofuran-1(3H)-one (2).

Keywords: Helichrysum arenarium (L.); Compositae; naphthoic acid; isobenzofuran

## 1. Introduction

*Helichrysum arenarium* (L.) Moench (Compositae) is distributed mainly in Xinjiang Uygur Autonomous Region of China, Mongolia, Europe, and Russia. It is a medicinal plant that has been long known in European traditional medicine as a cholagogue, choleretic, hepatoprotective, and detoxifying agents [1,2]. It also has activities of antioxidant, anti-bacteria on the basis of modern pharmacological research [3,4], and previous phytochemical study on this plant revealed the presence of flavonoids and phenolic acids [1]. In this paper, we report the isolation and structural elucidation of two new compounds from the titled whole herb.

#### 2. Results and discussion

Compound **1** was obtained as white needles, mp 182–184°C. It showed positive reactions with FeCl<sub>3</sub> and bromocresol green reagent. The quasi-molecular ion at m/z 271.0578 [M+Na]<sup>+</sup> in the HR-ESI-MS spectrum indicated the molecular formula to be C<sub>13</sub>H<sub>12</sub>O<sub>5</sub>. The <sup>1</sup>H NMR spectrum of **1**  showed the presence of two active proton singlets at  $\delta$  12.46 (1H, br s), 9.13 (1H, br s), two aromatic proton singlets at  $\delta$  7.19 (1H, s), 7.42 (1H, s), two ortho-coupled proton signals at  $\delta$  6.84 (1H, d, J = 8.4 Hz), 7.43 (1H, d, J = 8.4 Hz), and two methoxyl groups at  $\delta$  3.79 (6H, s,  $-\text{OCH}_3 \times 2$ ). The <sup>13</sup>C NMR spectrum of 1 had 13 carbon signals among which, except for one carboxyl carbon at  $\delta$  167.2 and two methoxyl carbons at  $\delta$  55.6, 56.1 mentioned above, 10 aromatic carbons were observed at δ 151.1, 147.5, 147.3, 140.3, 123.6, 121.7, 120.5, 115.1, 112.8, and 106.9, which were considered to be a naphthalene nucleus. By analysis of the <sup>13</sup>C NMR spectral data [5], the signals at  $\delta$  147.5 and 147.3 were two  $\beta$  carbon *ortho*-substituted by two methoxyl groups, which had correlations with two aromatic proton singlets at  $\delta$  7.19 (H-5) and 7.42 (H-8) in the NOESY spectrum of **1** (Figure 2). The signals at  $\delta$  151.1 and 140.3 were suggested to be two  $\alpha$ -C (C-1 and C-4) connected the carboxyl group and the hydroxyl group, respectively, based on the <sup>1</sup>H NMR spectral data and the HMBC correlations (Table 1). The locations

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Compound 1         Compound 2 $2$ $\delta_{\rm H}$ HMBC         NOESY         No. $\delta_{\rm C}$ $\delta_{\rm H}$ HMBC         NOESY $3$ $6$ $7.43$ (H, d, $8.4$ Hz) $-COOH$ , $4^{\rm C}$ $3.4$ $5.16$ (2H, s) $1.4$ , $7a^{\rm C}$ $1.4$ $3$ $6.44$ (H, d, $8.4$ Hz) $-COOH$ , $4^{\rm C}$ $3.4$ $112.7$ $5.16$ (2H, s) $1.4$ , $7a^{\rm C}$ $1.4$ $3$ $7.42$ (H, s) $4.7$ $2.4$ $3.199$ $6.50$ (H, s) $1.4$ , $7a^{\rm C}$ $1.4$ $3$ $7.19$ (H, s) $4.7$ $2.4$ $112.7$ $7.0$ CH <sub>3</sub> $5.16$ (2H, s) $1.4$ $1.7$ $1.4$ $3$ $7.19$ (H, s) $4.7$ $2.4$ $1.12.7$ $5.64$ $4.4$ , $5.6$ $7.0$ CH <sub>3</sub> $3$ $7.19$ (H, s) $1.5$ $5.33$ (H, t, $6.9$ Hz) $4.4$ , $4.5^{\rm C}$ $1.5^{\rm H}$ $3$ $7.19$ (Hr s) $1.7$ $5.33$ (H, $1.60$ Hz) $2.4^{\rm H}$ $5.4^{\rm H}$ $1.6^{\rm H}$ $3$ $3.79$ (3H, s) <th>ar</th> <th>nd <sup>13</sup>C NM</th> <th>R spectral data of compo-</th> <th>unds 1 and 2 (300)</th> <th>MHz for <sup>1</sup>H</th> <th>and 75 MHz</th> <th>for <sup>13</sup>C in I</th> <th><math>OMSO-d_6</math>, <math>\delta</math> values).</th> <th></th> <th></th>	ar	nd <sup>13</sup> C NM	R spectral data of compo-	unds 1 and 2 (300)	MHz for <sup>1</sup> H	and 75 MHz	for <sup>13</sup> C in I	$OMSO-d_6$ , $\delta$ values).		
			Compound 1					Compound 2		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\delta_{\mathrm{C}}$		$\delta_{\rm H}$	HMBC	NOESY	No.	$\delta_{\rm C}$	$\delta_{\rm H}$	HMBC	NOESY
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	140.3						168.2			
	123.6		7.43 (1H, d, 8.4 Hz)	-COOH, 4-C	3-H	3	67.4	5.16 (2H, s)	1, 4, 7a-C	1'-H
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	115.1		6.84 (1H, d, 8.4 Hz)	4a-C	2-H	3a	149.9			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	151.1					4	112.7			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$121.7^{a}$					5	162.0			
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	106.9		7.42 (1H, s)	4-C		6	98.9	6.50 (1H, s)	4, 7a-C	7-OCH <sub>3</sub>
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$147.5^{a}$					7	157.5			
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	147.3°	_				7a	103.4			
a 12.46 (br s) 12.46 (br s) 9.13 (br s) 9.14 (1.76 (3H, s)) 7.C $5-H$ 9.14 (1.70 (3H, s)) 7.5 (2', 4'-H) 1'' (102.6 4, 12 (1H, d, 7.5 Hz)) 7.4'-C $2', 4'-H$ 7.4'-C $2', 4'-H$ 7.5'-C $2', 4'-$	112.8		7.19 (1H, s)	1-C		1'	23.5	3.20 (2H, d, 6.9 Hz)	3a, 5, 3'-C	3, 2′, 4′-H
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	120.5	-				2'	125.7	5.33 (1H, t, 6.9 Hz)	4, 4', 5'-C	1′, 5′-H
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	167.2		12.46 (br s)			3/	132.1			
<sup>a</sup> 3.79 (3H, s) 6-C 5-H 5' 21.4 1.70 (3H, s) 2', 4'-C 2', 4'-H <sup>a</sup> 3.79 (3H, s) 7-C 8-H 1'' 102.6 4.12 (1H, d, 7.5Hz) 4'-C 2', 4'-H 2'' 73.3 76.7 4.12 (1H, d, 7.5Hz) 4'-C 2', 4'-H $3'' 76.7 5.6 69.7 5'' 65.8 10.62 (1H, br s) 7-C 6-H$			9.13 (br s)			4	66.4	4.23 (2H, s)	2′, 5′-C	1′, 5′-H
<sup>a</sup> 3.79 (3H, s) 7-C 8-H 1 <sup>"</sup> 102.6 4.12 (1H, d, 7.5 Hz) 4'-C 2 <sup>"</sup> 73.3 7.67 3" 76.7 4" 69.7 5-OH 10.62 (1H, br s) 7-OCH <sub>3</sub> 55.5 3.80 (3H, s) 7-C 6-H	56.1	а	3.79 (3H, s)	6-C	5-H	5'	21.4	1.70 (3H, s)	2′, 4′-C	2′, 4′-H
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	55.6	8	3.79 (3H, s)	7-C	8-H	$1^{\prime\prime}$	102.6	4.12 (1H, d, 7.5 Hz)	4'-C	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						2"	73.3			
$\begin{array}{cccc} 4'' & 69.7 \\ 5'' & 65.8 \\ 5-OH & 10.62 (1H, \rm br~s) \\ 7-\rm OCH_3 & 55.5 & 3.80 (3H, \rm s) & 7-C & 6-H \end{array}$						3//	76.7			
5" 65.8 5-OH 10.62 (1H, br s) 7-OCH <sub>3</sub> 55.5 3.80 (3H, s) 7-C 6-H						4″	69.7			
5-OH $10.62 (1H, br s)$ 7-OCH <sub>3</sub> 55.5 $3.80 (3H, s)$ 7-C $6-H$						5"	65.8			
7-0CH <sub>3</sub> 55.5 3.80 (3H, s) 7-C 6-H						5-OH		10.62 (1H, br s)		
						7-OCH <sub>3</sub>	55.5	3.80 (3H, s)	7-C	H-9

<sup>a</sup> Data may be exchangeable.

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Figure 1. Structures of compounds 1 and 2.

of 1-COOH and 4-OH were confirmed due to the signals at  $\delta$  7.43 (H-2) had long-range correlations with —COOH and C-4 in the HMBC experiment (Figure 2). The HMBC spectrum also showed the long-range correlations between H-3 and C-4<sub>a</sub>, H-5 and C-4, and H-8 and C-1. With the above data, the structure of **1** was established as 6,7-dimethoxy-4-hydroxy-1-naphthoic acid (Figure 1).

Compound **2** was obtained as white needles, mp  $175-177^{\circ}$ C. It showed positive reactions with FeCl<sub>3</sub> and the Molish reagent. The sugar was identified as xylose by acid hydrolysis and co-TLC with an authentic sample. The molecular formula was analyzed as C<sub>19</sub>H<sub>24</sub>O<sub>9</sub> from its quasi-molecular ion at

m/z 419.1315  $[M+Na]^+$  in the HR-ESI-MS spectrum and by <sup>1</sup>H and <sup>13</sup>C NMR spectral data. The IR spectrum of 2 indicated the presence of hydroxyl  $(3397 \text{ cm}^{-1})$ , ester carbonyl  $(1702 \text{ cm}^{-1})$ , and benzene ring (1602 and  $1533 \text{ cm}^{-1}$ ). <sup>1</sup>H NMR spectrum of **2** showed an active proton at  $\delta$  10.62 (1H, br s), an aromatic proton at  $\delta$  6.50 (1H, s), a methoxyl group at  $\delta$  3.80 (3H, s), a methyl group at  $\delta$  1.70 (3H, s), two oxygen-bearing methylene singlets at  $\delta$  4.23 (2H, s), 5.16 (2H, s), and a coupled group of ortho-coupled proton signals at  $\delta$  5.33 (1H, t, J = 6.9 Hz), 3.20 (2H, d, J = 6.9 Hz). Additionally, the anomeric proton of xylose was observed at  $\delta$  4.12 (1H, d, J = 7.5), so the anomeric configuration of the xylose was  $\beta$ -orientation.



Figure 2. Key HMBC and NOESY correlations of compounds 1 and 2.

In the  ${}^{13}C$  NMR spectrum of **2** exhibited 19 carbon signals, including a methoxy group  $(\delta 55.5)$ , a ester carbonyl ( $\delta 168.2$ ), two oxygen-bearing methylene ( $\delta$  66.4, 67.4), a methylene ( $\delta$  23.5), six aromatic and two double bond carbons at  $\delta$  149.9, 112.7, 162.0, 98.9, 157.5, 103.4, 132.1, 125.7, and a methyl group ( $\delta$  21.4) except for a set of xylosyl group, their corresponding carbon signals were assigned due to the HMBC correlations (Table 1). In the HMBC experiment (Figure 2), the long-range correlations between the anomeric proton of xylose at  $\delta$  4.12 (H-1") and C-4', H-4' and C-2', H-5' and C-2', and H-1' and C-3' were observed. It displayed the presence of isopentenyl-connected xylose moiety in structure 2. Furthermore, the configuration of the isopentenyl group was identified as Z due to the NOE correlations between H-5' and H-2', H-1' and H-4' in the NOESY spectrum (Figure 2).

The remaining aromatic carbon signals at δ 162.0 (C-5), 157.5 (C-7), 149.9 (C-3a), 112.7 (C-4), 103.4 (C-7a), and 98.9 (C-6) should be five-substituted benzene ring based on the <sup>1</sup>H NMR spectral data. The HMBC correlations between the proton signal at  $\delta$  3.80 (-OCH<sub>3</sub>) and C-7, H-6 and C-4, and H-1' and C-5 were observed (Figure 2). Remaining ester carbonyl ( $\delta$  168.2) and an oxygen-bearing methylene should form a five-numbered lactone, which connected C-3a and 7a of the benzene ring due to the long-range correlations of oxygen-bearing methylene signal at  $\delta$  5.16 (H-3) with C-4, C-7a, and C-1. Therefore, the structure of 2 was established as (Z)-5-hydroxy-7-methoxy-4-[3-methyl-4-(O-β-D-xylopyranosyl)but-2enyl]isobenzofuran-1(3H)-one (Figure 1).

#### 3. Experimental

#### 3.1 General experimental procedures

Melting points were determined on Yanaco MP-S3 melting point apparatus and are uncorrected. The optical rotation was measured on Perkin-Elmer 241 polarimeter. The UV spectra were recorded on a Shimadzu UV-260 UV-VIS instrument. IR spectra were recorded on a Bruker IFS-55 instrument as KBr pellets. NMR spectra were recorded with a Bruker ARX-300 or Bruker ARX-600. ESI-MS was performed on Finnigan LCQ mass spectrometer. HR-ESI-MS data were performed on QSTARLCQ mass spectrometer.

#### 3.2 Plant material

The plant material of *H. arenarium* (L.) Moench was collected in August 2006, in Xinjiang Province, China, and identified by Prof. Qishi Sun (Drug Inspection Institute). A voucher specimen (No. 20060820) is deposited in Research Department of Natural Medicine, Shenyang Pharmaceutical University.

#### 3.3 Extraction and isolation

Dried whole plant (6.5 kg) of *H. arenarium* was extracted with 70% ethanol. The extract was concentrated *in vacuo*, then partitioned with petroleum ether, CHCl<sub>3</sub>, EtOAc, and *n*-BuOH successively. The EtOAc extract (28 g) was subjected to column chromatography on silica gel gradiently eluted with CHCl<sub>3</sub>–MeOH to give fraction 3 (100:1) and fraction 13 (100:6). Fraction 3 was rechromatographed on silica gel gradiently eluted with CHCl<sub>3</sub>–EtOAc to give subfraction 4 (100:7). Subfractions 4 and 13 were purified on Sephadex LH-20 with CHCl<sub>3</sub>–MeOH (1:1) to yield **1** (31.0 mg) and **2** (42.0 mg), respectively.

## 3.3.1 Compound 1

White needles (MeOH); mp 182–184°C; UV (CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\varepsilon$ ): 250 (2.40), 295 (3.22) nm; IR (KBr)  $\nu_{max}$  (cm<sup>-1</sup>): 3428, 2940, 1680, 1597, 1524, 1432, 1283, 1206, 1117, 763; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; HR-ESI-MS: *m/z* 271.0578 (calcd for C<sub>13</sub>H<sub>12</sub>O<sub>5</sub>Na, 271.0582); ESI-MS *m/z* 271 [M+Na]<sup>+</sup>.

## 3.3.2 Compound 2

White needles (CHCl<sub>3</sub>–MeOH); mp 175– 177°C;  $[\alpha]_D^{20} - 5.777 \ (c = 0.1, CH_3CH_2OH);$ UV (CH<sub>3</sub>CH<sub>2</sub>OH)  $\lambda_{max} (\log \varepsilon)$ : 226 (2.00), 260 (0.78), 296 (0.40) nm; IR (KBr)  $\nu_{max} \ (cm^{-1})$ : 3397, 2917, 1738, 1702, 1602, 1533, 1448, 1356, 1226, 1060, 839; <sup>1</sup>H and <sup>13</sup>C NMR spectral data, see Table 1; HR-ESI-MS: *m/z* 419.1315 (calcd for C<sub>19</sub>H<sub>24</sub>O<sub>9</sub>Na, 419.1318); ESI-MS *m/z* 419 [M+Na]<sup>+</sup>, 815 [2M+Na]<sup>+</sup>, 395 [M-H]<sup>-</sup>, 791 [2M-H]<sup>-</sup>.

# 3.4 Acid hydrolysis of compound 2

Compound **2** (5 mg) was refluxed in a mixture of concentrated HCl-H<sub>2</sub>O-EtOH (2:1:2) (10 ml) on a water bath for 2 h. The hydrolysate was partitioned between EtOAc and H<sub>2</sub>O, and the aqueous layer was compared with authentic samples on TLC with silica gel [CHCl<sub>3</sub>-  $CH_3OH-H_2O$  (9:6:1), lower phase], which showed the sugar was xylose.

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